



## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

CHEMetrics, Inc.

SDS No: R5510

Version No: 5.11

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 4

Issue Date: 13/08/2018

Print Date: 13/08/2018

S.GHS.USA.EN

### SECTION 1 IDENTIFICATION

#### Product Identifier

Product name	Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill
Synonyms	Part Nos.: R-5510, R-5510A, R-5510B, R-5510C, R-5510D, K-5543 Ampoules, R-7870
Proper shipping name	Corrosive liquids, n.o.s. (contains sulfuric acid and acetic acid glacial)
Other means of identification	Not Available

#### Recommended use of the chemical and restrictions on use

Relevant identified uses	Component of water analysis test kits K-5510, K-5510A, K-5510B, K-5510C, K-5510D, K-5543, K-7870
--------------------------	--

#### Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	CHEMetrics, Inc.
Address	4295 Catlett Road, Midland VA - 22728 United States
Telephone	1-540-788-9026
Fax	1-540-788-4856
Website	www.chemetrics.com
Email	technical@chemetrics.com

#### Emergency phone number

Association / Organisation	ChemTel, Inc.
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	+01-813-248-0585

### SECTION 2 HAZARD(S) IDENTIFICATION

#### Classification of the substance or mixture

Classification	Metal Corrosion Category 1, Skin Corrosion/Irritation Category 1A, Serious Eye Damage Category 1, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation)
----------------	--

#### Label elements

Hazard pictogram(s)	 
SIGNAL WORD	<b>DANGER</b>

#### Hazard statement(s)

H290	May be corrosive to metals.
H314	Causes severe skin burns and eye damage.
H335	May cause respiratory irritation.

#### Hazard(s) not otherwise specified

Not Applicable

#### Precautionary statement(s) General

P101	If medical advice is needed, have product container or label at hand.
P102	Keep out of reach of children.

Continued...

# Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

**P103** Read label before use.

## Precautionary statement(s) Prevention

<b>P260</b>	Do not breathe dust/fume/gas/mist/vapours/spray.
<b>P271</b>	Use only outdoors or in a well-ventilated area.
<b>P280</b>	Wear protective gloves/protective clothing/eye protection/face protection.
<b>P234</b>	Keep only in original container.

## Precautionary statement(s) Response

<b>P301+P330+P331</b>	IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
<b>P303+P361+P353</b>	IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower.
<b>P305+P351+P338</b>	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
<b>P310</b>	Immediately call a POISON CENTER or doctor/physician.
<b>P363</b>	Wash contaminated clothing before reuse.
<b>P390</b>	Absorb spillage to prevent material damage.
<b>P304+P340</b>	IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing.

## Precautionary statement(s) Storage

<b>P405</b>	Store locked up.
<b>P403+P233</b>	Store in a well-ventilated place. Keep container tightly closed.

## Precautionary statement(s) Disposal

<b>P501</b>	Dispose of contents/container in accordance with local regulations.
-------------	---

## SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

### Substances

See section below for composition of Mixtures

### Mixtures

CAS No	%[weight]	Name
7732-18-5	>88	<u>water</u>
64-19-7	5	<u>acetic acid glacial</u>
1762-95-4	3	<u>ammonium thiocyanate</u>
67-56-1	<1	<u>methanol</u>
7783-85-9	<1	<u>ammonium ferrous sulfate</u>
7664-93-9	<1	<u>sulfuric acid</u>
497-18-7	<0.01	<u>carbohydrazide</u>

## SECTION 4 FIRST-AID MEASURES

### Description of first aid measures

<b>Eye Contact</b>	If this product comes in contact with the eyes: <ul style="list-style-type: none"> <li>► Immediately hold eyelids apart and flush the eye continuously with running water.</li> <li>► Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>► Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.</li> <li>► Transport to hospital or doctor without delay.</li> <li>► Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
	If skin or hair contact occurs: <ul style="list-style-type: none"> <li>► Immediately flush body and clothes with large amounts of water, using safety shower if available.</li> <li>► Quickly remove all contaminated clothing, including footwear.</li> <li>► Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.</li> <li>► Transport to hospital, or doctor.</li> </ul>
<b>Inhalation</b>	<ul style="list-style-type: none"> <li>► If fumes or combustion products are inhaled remove from contaminated area.</li> <li>► Lay patient down. Keep warm and rested.</li> <li>► Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>► Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>► Transport to hospital, or doctor, without delay.</li> <li>► Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema.</li> <li>► Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).</li> <li>► As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested.</li> <li>► Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered.</li> </ul> <p><b>This must definitely be left to a doctor or person authorised by him/her.</b></p> <p>(ICSC13719)</p>

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

### Ingestion

- ▶ For advice, contact a Poisons Information Centre or a doctor at once.
- ▶ Urgent hospital treatment is likely to be needed.
- ▶ **If swallowed do NOT induce vomiting.**
- ▶ If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.
- ▶ Observe the patient carefully.
- ▶ Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.
- ▶ Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.
- ▶ Transport to hospital or doctor without delay.

### Most important symptoms and effects, both acute and delayed

See Section 11

### Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

For acute or short term repeated exposures to strong acids:

- ▶ Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- ▶ Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling.
- ▶ Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- ▶ Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

#### INGESTION:

- ▶ Immediate dilution (milk or water) within 30 minutes post ingestion is recommended.
- ▶ **DO NOT attempt to neutralise the acid since exothermic reaction may extend the corrosive injury.**
- ▶ Be careful to avoid further vomit since re-exposure of the mucosa to the acid is harmful. Limit fluids to one or two glasses in an adult.
- ▶ Charcoal has no place in acid management.
- ▶ Some authors suggest the use of lavage within 1 hour of ingestion.

#### SKIN:

- ▶ Skin lesions require copious saline irrigation. Treat chemical burns as thermal burns with non-adherent gauze and wrapping.
- ▶ Deep second-degree burns may benefit from topical silver sulfadiazine.

#### EYE:

- ▶ Eye injuries require retraction of the eyelids to ensure thorough irrigation of the conjunctival cul-de-sacs. Irrigation should last at least 20-30 minutes. **DO NOT use neutralising agents or any other additives.** Several litres of saline are required.
- ▶ Cycloplegic drops, (1% cyclopentolate for short-term use or 5% homatropine for longer term use) antibiotic drops, vasoconstrictive agents or artificial tears may be indicated dependent on the severity of the injury.
- ▶ Steroid eye drops should only be administered with the approval of a consulting ophthalmologist.

[Ellenhorn and Barceloux: Medical Toxicology]

For acute and short term repeated exposures to methanol:

- ▶ Toxicity results from accumulation of formaldehyde/formic acid.
- ▶ Clinical signs are usually limited to CNS, eyes and GI tract. Severe metabolic acidosis may produce dyspnea and profound systemic effects which may become intractable. All symptomatic patients should have arterial pH measured. Evaluate airway, breathing and circulation.
- ▶ Stabilise obtunded patients by giving naloxone, glucose and thiamine.
- ▶ Decontaminate with Ipecac or lavage for patients presenting 2 hours post-ingestion. Charcoal does not absorb well; the usefulness of cathartic is not established.
- ▶ Forced diuresis is not effective; haemodialysis is recommended where peak methanol levels exceed 50 mg/dL (this correlates with serum bicarbonate levels below 18 meq/L).
- ▶ Ethanol, maintained at levels between 100 and 150 mg/dL, inhibits formation of toxic metabolites and may be indicated when peak methanol levels exceed 20 mg/dL. An intravenous solution of ethanol in D5W is optimal.
- ▶ Folate, as leucovorin, may increase the oxidative removal of formic acid. 4-methylpyrazole may be an effective adjunct in the treatment. Phenytoin may be preferable to diazepam for controlling seizure.

[Ellenhorn Barceloux: Medical Toxicology]

### BIOLOGICAL EXPOSURE INDEX - BEI

Determinant	Index	Sampling Time	Comment
1. Methanol in urine	15 mg/l	End of shift	B, NS
2. Formic acid in urine	80 mg/gm creatinine	Before the shift at end of workweek	B, NS

B: Background levels occur in specimens collected from subjects **NOT** exposed.

NS: Non-specific determinant - observed following exposure to other materials.

For thiocyanate poisonings haemodialysis is recommended as the treatment of choice. Phenobarbital protects poisoned animals against death. Thiocyanate ion is slowly excreted in the urine and is not decomposed to any appreciable degree to cyanide.

[GOSSELIN, SMITH & HODGE: Clinical Toxicology of Commercial Products 5th Ed]

## SECTION 5 FIRE-FIGHTING MEASURES

### Extinguishing media

- ▶ Water spray or fog.
- ▶ Foam.
- ▶ Dry chemical powder.
- ▶ BCF (where regulations permit).
- ▶ Carbon dioxide.

### Special hazards arising from the substrate or mixture

Fire Incompatibility	None known.
----------------------	-------------

### Special protective equipment and precautions for fire-fighters

#### Fire Fighting

- ▶ Alert Fire Brigade and tell them location and nature of hazard.
- ▶ Wear full body protective clothing with breathing apparatus.
- ▶ Prevent, by any means available, spillage from entering drains or water course.
- ▶ Use fire fighting procedures suitable for surrounding area.
- ▶ **Do not approach containers suspected to be hot.**
- ▶ Cool fire exposed containers with water spray from a protected location.
- ▶ If safe to do so, remove containers from path of fire.

Continued...

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

<b>Fire/Explosion Hazard</b>	<ul style="list-style-type: none"> <li>► Equipment should be thoroughly decontaminated after use.</li> <li>► Non combustible.</li> <li>► Not considered to be a significant fire risk.</li> <li>► Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.</li> <li>► Heating may cause expansion or decomposition leading to violent rupture of containers.</li> <li>► May emit corrosive, poisonous fumes. May emit acrid smoke.</li> </ul>
------------------------------	---

## SECTION 6 ACCIDENTAL RELEASE MEASURES

### Personal precautions, protective equipment and emergency procedures

See section 8

### Environmental precautions

See section 12

### Methods and material for containment and cleaning up

<b>Minor Spills</b>	<ul style="list-style-type: none"> <li>► Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.</li> <li>► Check regularly for spills and leaks.</li> <li>► Clean up all spills immediately.</li> <li>► Avoid breathing vapours and contact with skin and eyes.</li> <li>► Control personal contact with the substance, by using protective equipment.</li> <li>► Contain and absorb spill with sand, earth, inert material or vermiculite.</li> <li>► Wipe up.</li> <li>► Place in a suitable, labelled container for waste disposal.</li> </ul>
<b>Major Spills</b>	<ul style="list-style-type: none"> <li>► Clear area of personnel and move upwind.</li> <li>► Alert Fire Brigade and tell them location and nature of hazard.</li> <li>► Wear full body protective clothing with breathing apparatus.</li> <li>► Prevent, by any means available, spillage from entering drains or water course.</li> <li>► Consider evacuation (or protect in place).</li> <li>► Stop leak if safe to do so.</li> <li>► Contain spill with sand, earth or vermiculite.</li> <li>► Collect recoverable product into labelled containers for recycling.</li> <li>► Neutralise/decontaminate residue (see Section 13 for specific agent).</li> <li>► Collect solid residues and seal in labelled drums for disposal.</li> <li>► Wash area and prevent runoff into drains.</li> <li>► After clean up operations, decontaminate and launder all protective clothing and equipment before storing and re-using.</li> <li>► If contamination of drains or waterways occurs, advise emergency services.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

## SECTION 7 HANDLING AND STORAGE

### Precautions for safe handling

<b>Safe handling</b>	<ul style="list-style-type: none"> <li>► Avoid all personal contact, including inhalation.</li> <li>► Wear protective clothing when risk of exposure occurs.</li> <li>► Use in a well-ventilated area.</li> <li>► <b>WARNING:</b> To avoid violent reaction, <b>ALWAYS</b> add material to water and <b>NEVER</b> water to material.</li> <li>► Avoid smoking, naked lights or ignition sources.</li> <li>► Avoid contact with incompatible materials.</li> <li>► When handling, <b>DO NOT eat, drink or smoke.</b></li> <li>► Keep containers securely sealed when not in use.</li> <li>► Avoid physical damage to containers.</li> <li>► Always wash hands with soap and water after handling.</li> <li>► Work clothes should be laundered separately. Launder contaminated clothing before re-use.</li> <li>► Use good occupational work practice.</li> <li>► Observe manufacturer's storage and handling recommendations contained within this SDS.</li> <li>► Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions are maintained.</li> </ul> <p><b>Wear impact- and splash-resistant eyewear. Break the ampoule tip only when it is completely immersed in sample. Breaking the tip in air may cause the glass ampoule to shatter.</b></p>
<b>Other information</b>	<ul style="list-style-type: none"> <li>► Store in original containers.</li> <li>► Keep containers securely sealed.</li> <li>► Store in a cool, dry, well-ventilated area.</li> <li>► Store away from incompatible materials and foodstuff containers.</li> <li>► Protect containers against physical damage and check regularly for leaks.</li> <li>► Observe manufacturer's storage and handling recommendations contained within this SDS.</li> </ul> <p><b>For optimum analytical performance, store in the dark and at room temperature.</b></p>

### Conditions for safe storage, including any incompatibilities

<b>Suitable container</b>	<ul style="list-style-type: none"> <li>► <b>DO NOT</b> use aluminium or galvanised containers</li> <li>► Check regularly for spills and leaks</li> <li>► Lined metal can, lined metal pail/ can.</li> <li>► Plastic pail.</li> <li>► Polyliner drum.</li> <li>► Packing as recommended by manufacturer.</li> <li>► Check all containers are clearly labelled and free from leaks.</li> </ul> <p>For low viscosity materials</p> <ul style="list-style-type: none"> <li>► Drums and jerricans must be of the non-removable head type.</li> <li>► Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> </ul> <p>For materials with a viscosity of at least 2680 cSt. (23 deg. C) and solids (between 15 C deg. and 40 deg C.):</p> <ul style="list-style-type: none"> <li>► Removable head packaging;</li> </ul>
---------------------------	--

Continued...

## Hydrogen Peroxide CHEMets &amp; VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

	<ul style="list-style-type: none"> <li>▶ Cans with friction closures and</li> <li>▶ low pressure tubes and cartridges</li> </ul> <p>may be used.</p> <p>-</p> <p>Where combination packages are used, and the inner packages are of glass, porcelain or stoneware, there must be sufficient inert cushioning material in contact with inner and outer packages unless the outer packaging is a close fitting moulded plastic box and the substances are not incompatible with the plastic.</p>
Storage incompatibility	<ul style="list-style-type: none"> <li>▶ Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.</li> <li>▶ Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts - neutralisation can generate dangerously large amounts of heat in small spaces.</li> <li>▶ The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.</li> <li>▶ The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can sputter the acid.</li> <li>▶ Inorganic acids react with active metals, including such structural metals as aluminum and iron, to release hydrogen, a flammable gas.</li> <li>▶ Inorganic acids can initiate the polymerisation of certain classes of organic compounds.</li> <li>▶ Inorganic acids react with cyanide compounds to release gaseous hydrogen cyanide.</li> <li>▶ Inorganic acids generate flammable and/or toxic gases in contact with dithiocarbamates, isocyanates, mercaptans, nitrides, nitriles, sulfides, and strong reducing agents. Additional gas-generating reactions occur with sulfites, nitrites, thiosulfates (to give H<sub>2</sub>S and SO<sub>3</sub>), dithionites (SO<sub>2</sub>), and even carbonates.</li> <li>▶ Acids often catalyse (increase the rate of) chemical reactions.</li> <li>▶ Reacts with mild steel, galvanised steel / zinc producing hydrogen gas which may form an explosive mixture with air.</li> </ul> <p>Acetic acid:</p> <ul style="list-style-type: none"> <li>▶ vapours forms explosive mixtures with air (above 39 C.)</li> <li>▶ reacts violently with bases such as carbonates and hydroxides (giving off large quantities of heat), oxidisers, organic amines, acetaldehyde, potassium tert-butoxide</li> <li>▶ reacts (sometimes violently), with strong acids, aliphatic amines, alkanolamines, alkylene oxides, epichlorohydrin, acetic anhydride, 2-aminoethanol, ammonia, ammonium nitrate, bromine pentafluoride, chlorosulfonic acid, chromic acid, chromium trioxide, ethylenediamine, ethyleneimine, hydrogen peroxide, isocyanates, oleum, perchloric acid, permanganates, phosphorus isocyanate, phosphorus trichloride, sodium peroxide, xylene</li> <li>▶ attacks cast iron, stainless steel and other metals, forming flammable hydrogen gas</li> <li>▶ attacks many forms of rubber, plastics and coatings</li> </ul>

## SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

## Control parameters

## OCCUPATIONAL EXPOSURE LIMITS (OEL)

## INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US NIOSH Recommended Exposure Limits (RELs)	acetic acid glacial	Acetic acid (aqueous), Ethanoic acid, Glacial acetic acid (pure compound), Methanecarboxylic acid [Note: Can be found in concentrations of 5-8% in vinegar.]	10 ppm / 25 mg/m3	37 mg/m3 / 15 ppm	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	acetic acid glacial	Acetic acid	10 ppm	15 ppm	Not Available	TLV® Basis: URT & eye irr; pulm func
US OSHA Permissible Exposure Levels (PELs) - Table Z1	acetic acid glacial	Acetic acid	10 ppm / 25 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	methanol	Carbinol, Columbian spirits, Methanol, Pyroligneous spirit, Wood alcohol, Wood naphtha, Wood spirit	200 ppm / 260 mg/m3	325 mg/m3 / 250 ppm	Not Available	[skin]
US ACGIH Threshold Limit Values (TLV)	methanol	Methanol	200 ppm	250 ppm	Not Available	TLV® Basis: Headache; eye dam; dizziness; nausea; BEI
US OSHA Permissible Exposure Levels (PELs) - Table Z1	methanol	Methyl alcohol	200 ppm / 260 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	ammonium ferrous sulfate	Iron salts, soluble, as Fe	1 mg/m3	Not Available	Not Available	TLV® Basis: URT & skin irr
US NIOSH Recommended Exposure Limits (RELs)	sulfuric acid	Battery acid, Hydrogen sulfate, Oil of vitriol, Sulfuric acid (aqueous)	1 mg/m3	Not Available	Not Available	Not Available
US ACGIH Threshold Limit Values (TLV)	sulfuric acid	Sulfuric acid	0.2 mg/m3	Not Available	Not Available	TLV® Basis: Pulm func
US OSHA Permissible Exposure Levels (PELs) - Table Z1	sulfuric acid	Sulfuric acid	1 mg/m3	Not Available	Not Available	Not Available

## EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
acetic acid glacial	Acetic acid	Not Available	Not Available	Not Available
ammonium thiocyanate	Ammonium thiocyanate	2.3 mg/m3	25 mg/m3	150 mg/m3
methanol	Methyl alcohol; (Methanol)	Not Available	Not Available	Not Available
ammonium ferrous sulfate	Ferrous ammonium sulfate	15 mg/m3	170 mg/m3	1,000 mg/m3
ammonium ferrous sulfate	Ferric ammonium sulfate; (Sulfuric acid, ammonium iron(3e+) salt (2:1:1))	14 mg/m3	160 mg/m3	950 mg/m3
ammonium ferrous sulfate	Ferric ammonium sulfate dodecahydrate	26 mg/m3	280 mg/m3	1,700 mg/m3
ammonium ferrous sulfate	Ammonium ferrous sulfate hexahydrate	9.6 mg/m3	110 mg/m3	640 mg/m3
sulfuric acid	Sulfuric acid	Not Available	Not Available	Not Available

Ingredient	Original IDLH	Revised IDLH
------------	---------------	--------------

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

water	Not Available	Not Available
acetic acid glacial	50 ppm	Not Available
ammonium thiocyanate	Not Available	Not Available
methanol	6,000 ppm	Not Available
ammonium ferrous sulfate	Not Available	Not Available
sulfuric acid	15 mg/m <sup>3</sup>	Not Available
carbohydrazide	Not Available	Not Available

### Exposure controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection.

An approved self contained breathing apparatus (SCBA) may be required in some situations.

Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

Type of Contaminant:	Air Speed:
solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min.)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.



► Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure.

► Chemical goggles whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted.

► Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection.

► Alternatively a gas mask may replace splash goggles and face shields.

► Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

#### Personal protection



#### Eye and face protection

► Safety glasses with unperforated side shields may be used where continuous eye protection is desirable, as in laboratories; spectacles are not sufficient where complete eye protection is needed such as when handling bulk-quantities, where there is a danger of splashing, or if the material may be under pressure.

► Chemical goggles whenever there is a danger of the material coming in contact with the eyes; goggles must be properly fitted.

► Full face shield (20 cm, 8 in minimum) may be required for supplementary but never for primary protection of eyes; these afford face protection.

► Alternatively a gas mask may replace splash goggles and face shields.

► Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation - lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

#### Skin protection

See Hand protection below

► Elbow length PVC gloves  
► When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

#### Body protection

See Other protection below

► Overalls.  
► PVC Apron.  
► PVC protective suit may be required if exposure severe.  
► Eyewash unit.  
► Ensure there is ready access to a safety shower.

#### Other protection

## Hydrogen Peroxide CHEMets &amp; VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

## Recommended material(s)

## GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

**"Forsberg Clothing Performance Index".**

The effect(s) of the following substance(s) are taken into account in the **computer-generated** selection:

Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

Material	CPI
NEOPRENE	B
BUTYL	C
BUTYL/NEOPRENE	C
NAT+NEOPR+NITRILE	C
NATURAL RUBBER	C
NATURAL+NEOPRENE	C
NEOPRENE/NATURAL	C
NITRILE	C
NITRILE+PVC	C
PE	C
PE/EVAL/PE	C
PVA	C
PVC	C
PVDC/PE/PVDC	C
SARANEX-23 2-PLY	C
SARANEX-23	C
TEFLON	C
VITON	C
VITON/NEOPRENE	C

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

**NOTE:** As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

## Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS	-	A-PAPR-AUS / Class 1
up to 50 x ES	-	A-AUS / Class 1	-
up to 100 x ES	-	A-2	A-PAPR-2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO<sub>2</sub>), G = Agricultural chemicals, K = Ammonia(NH<sub>3</sub>), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

## SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

## Information on basic physical and chemical properties

Appearance	Colorless		
Physical state	Liquid	Relative density (Water = 1)	1.0
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	1.5	Decomposition temperature	Not Available
Melting point / freezing point (°C)	1	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	101	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

## SECTION 10 STABILITY AND REACTIVITY

Reactivity See section 7

Continued...

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

<b>Chemical stability</b>	<ul style="list-style-type: none"> <li>► Contact with alkaline material liberates heat</li> <li>► Unstable in the presence of incompatible materials.</li> <li>► Product is considered stable.</li> <li>► Hazardous polymerisation will not occur.</li> </ul>
<b>Possibility of hazardous reactions</b>	See section 7
<b>Conditions to avoid</b>	See section 7
<b>Incompatible materials</b>	See section 7
<b>Hazardous decomposition products</b>	See section 5

### SECTION 11 TOXICOLOGICAL INFORMATION

#### Information on toxicological effects

<b>Inhaled</b>	<p>The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. There may be dizziness, headache, nausea and weakness.</p> <p>Not normally a hazard due to non-volatile nature of product</p> <p>The material has <b>NOT</b> been classified by EC Directives or other classification systems as "harmful by inhalation". This is because of the lack of corroborating animal or human evidence.</p>
<b>Ingestion</b>	<p>Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident.</p> <p>The material has <b>NOT</b> been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence.</p>
<b>Skin Contact</b>	<p>Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue.</p> <p>Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions.</p> <p>Open cuts, abraded or irritated skin should not be exposed to this material</p> <p>Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.</p>
<b>Eye</b>	<p>If applied to the eyes, this material causes severe eye damage.</p> <p>Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns. Mild burns of the epithelia generally recover rapidly and completely.</p>
<b>Chronic</b>	<p>Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Irritation of airways to lung, with cough, and inflammation of lung tissue often occurs.</p> <p>Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems.</p> <p>Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.</p> <p>Repeated minor exposure to acetic acid by mouth can cause blackening of the skin and teeth, erosion of the teeth, vomiting, diarrhea and nausea. Repeated minor vapour exposure may cause chronic inflammation of the airways and bronchitis. Results from testing are mixed, with one report indicating only slight irritation to the airways, stomach and skin, while another reported inflammation of the conjunctiva, bronchi, pharynx and erosion of teeth. Exposure to higher levels caused blackening and hyperkeratosis of the skin and hands. Heartburn and constipation have also been reported with prolonged exposures.</p> <p>Long-term exposure to methanol vapour, at concentrations exceeding 3000 ppm, may produce cumulative effects characterised by gastrointestinal disturbances (nausea, vomiting), headache, ringing in the ears, insomnia, trembling, unsteady gait, vertigo, conjunctivitis and clouded or double vision. Liver and/or kidney injury may also result.</p>

<b>Hydrogen Peroxide CHEMets &amp; VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	Not Available	Not Available
<b>water</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	Not Available	Not Available
<b>acetic acid glacial</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	<p>Dermal (rabbit) LD50: 1060 mg/kg<sup>[2]</sup></p> <p>Inhalation (mammal) LC50: 11.4 mg/l/4h<sup>[2]</sup></p> <p>Oral (rat) LD50: 3310 mg/kg<sup>[2]</sup></p>	<p>Eye (rabbit): 0.05mg (open)-SEVERE</p> <p>Skin (human):50mg/24hr - mild</p> <p>Skin (rabbit):525mg (open)-SEVERE</p>
<b>ammonium thiocyanate</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	<p>dermal (rat) LD50: &gt;2000 mg/kg<sup>[1]</sup></p> <p>Oral (rat) LD50: 501 mg/kg<sup>[2]</sup></p>	<p>Not Available</p>
<b>methanol</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	<p>Dermal (rabbit) LD50: 15800 mg/kg<sup>[2]</sup></p> <p>Inhalation (rat) LC50: 63926.976 mg/l/4h<sup>[2]</sup></p> <p>Oral (rat) LD50: 5600 mg/kg<sup>[2]</sup></p>	<p>Eye (rabbit): 100 mg/24h-moderate</p> <p>Eye (rabbit): 40 mg-moderate</p> <p>Skin (rabbit): 20 mg/24 h-moderate</p>
<b>ammonium ferrous sulfate</b>	<b>TOXICITY</b>	<b>IRRITATION</b>
	Oral (rat) LD50: 3250 mg/kg <sup>[2]</sup>	Not Available

## Hydrogen Peroxide CHEMets &amp; VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

	TOXICITY	IRRITATION
sulfuric acid	Inhalation (guinea pig) LC50: 0.036 mg/l/8H <sup>[2]</sup> Oral (rat) LD50: 2140 mg/kg <sup>[2]</sup>	Eye (rabbit): 1.38 mg SEVERE Eye (rabbit): 5 mg/30sec SEVERE
carbohydrazide	TOXICITY Not Available	IRRITATION Not Available
<b>Legend:</b>	1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances	

WATER	No significant acute toxicological data identified in literature search.  The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis. The material may cause severe skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin. Repeated exposures may produce severe ulceration. Prolonged or repeated exposure to acetic acid may produce irritation and/or corrosion at the site of contact as well as systemic toxicity. Prolonged inhalation exposure results in muscle imbalance, increase in blood cholinesterase activity, decrease in albumin and decreased growth but no reproductive or foetal toxicity, according to animal testing.
ACETIC ACID GLACIAL	The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.  for hexahydrate RTECS No.: BR 6500000
METHANOL	
AMMONIUM FERROUS SULFATE	
SULFURIC ACID	<b>WARNING:</b> For inhalation exposure <u>ONLY</u> : This substance has been classified by the IARC as Group 1: <b>CARCINOGENIC TO HUMANS</b>  Occupational exposures to strong inorganic acid mists of sulfuric acid:  Convulsions and effects on monoamine oxidase recorded.
CARBOHYDRAZIDE	Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The disorder is characterized by difficulty breathing, cough and mucus production.
Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill & ACETIC ACID GLACIAL & AMMONIUM FERROUS SULFATE & SULFURIC ACID & CARBOHYDRAZIDE	For acid mists, aerosols, vapours  Test results suggest that eukaryotic cells are susceptible to genetic damage when the pH falls to about 6.5. Cells from the respiratory tract have not been examined in this respect. Mucus secretion may protect the cells of the airway from direct exposure to inhaled acidic mists (which also protects the stomach lining from the hydrochloric acid secreted there).

Acute Toxicity	🚫	Carcinogenicity	🚫
Skin Irritation/Corrosion	✓	Reproductivity	🚫
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	✓
Respiratory or Skin sensitisation	🚫	STOT - Repeated Exposure	🚫
Mutagenicity	🚫	Aspiration Hazard	🚫

Legend: ✗ – Data available but does not fill the criteria for classification  
✓ – Data available to make classification  
🚫 – Data Not Available to make classification

## SECTION 12 ECOLOGICAL INFORMATION

## Toxicity

Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	Not Available	Not Available	Not Available	Not Available	Not Available
water	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	Not Available	Not Available	Not Available	Not Available	Not Available
acetic acid glacial	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	31.3-67.6mg/L	2
	EC50	48	Crustacea	18.9mg/L	2
	EC50	72	Algae or other aquatic plants	29.23mg/L	2
ammonium thiocyanate	NOEC	48	Crustacea	21.5mg/L	2
	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	NOEC	504	Crustacea	1.25mg/L	2

Continued...

## Hydrogen Peroxide CHEMets &amp; VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
methanol	LC50	96	Fish	>100mg/L	4
	EC50	48	Crustacea	>10000mg/L	4
	EC50	96	Algae or other aquatic plants	<10000mg/L	4
	BCF	24	Algae or other aquatic plants	0.05mg/L	4
	EC0	168	Algae or other aquatic plants	=530mg/L	1
	NOEC	72	Crustacea	0.1mg/L	4
ammonium ferrous sulfate	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	39mg/L	4
	BCF	336	Algae or other aquatic plants	8mg/L	4
sulfuric acid	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	LC50	96	Fish	=8mg/L	1
	EC50	48	Crustacea	=42.5mg/L	1
	EC0	24	Crustacea	=30mg/L	1
carbohydrazide	ENDPOINT	TEST DURATION (HR)	SPECIES	VALUE	SOURCE
	EC50	72	Algae or other aquatic plants	0.46mg/L	2
	NOEC	72	Algae or other aquatic plants	0.074mg/L	2

## Legend:

Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

## Ecotoxicity:

The tolerance of water organisms towards pH margin and variation is diverse. Recommended pH values for test species listed in OECD guidelines are between 6.0 and almost 9. Acute testing with fish showed 96h-LC50 at about pH 3.5

## For Thiocyanates:

Terrestrial Fate: Soil - At near ambient temperatures (approximately 30 C.) it appears that sorption and volatilization are not significant processes for thiocyanate in soil, with thiocyanate losses due primarily to microbial degradation. Thiocyanate is not persistent in soils. Thiocyanate may undergo both aerobic and anaerobic microbial degradation; however, the degradation pathway has not been defined. Saturated soils treated with thiocyanate were found to emit carbonyl sulfide (COS). Microbial degradation is the primary mechanism for thiocyanate disappearance at or below 30 C., with carbonyl sulfide as a possible hydrolysis product. Loss of thiocyanate at higher temperatures (50-60 C.) did not appear to result from microbial degradation.

Aquatic Fate: Biodegradation is a significant transformation process for thiocyanates in natural waters. Thiocyanate is toxic to microorganisms at high concentrations; however, acclimated cultures have increased tolerance to this compound. At concentrations up to 1.42 g/L, thiocyanate is completely degraded within 4 days to ammonia and sulfate ion (SO4-2) by an acclimated co-culture of two bacteria (Acinetobacter johnsonii and Pseudomonas diminuta).

For Acetic Acid: Acetic acid and its salts (the acetates) can be grouped together because of their close structural relationships, their natural occurrence in plants and animals, and their fundamental role in cell metabolism.

Atmospheric Fate: Acetic acid is degraded photochemically in the atmosphere to produce hydroxyl radicals (estimated typical half-life of 22 days). Physical removal of acetates on atmospheric particulates may occur via wet or dry deposition.

Aquatic Fate: Natural water will neutralize dilute solutions of acetic acid. Spills of acetic acid on soil will readily biodegrade - the biodegradation rate for acetic acid after 14 days and under aerobic conditions is 74 days. Acetic acid is not expected to bioconcentrate in aquatic systems. Drinking water standards: none available.

Terrestrial Fate: Spills of acetic acid on soil will readily biodegrade - the biodegradation rate for acetic acid after 14 days under aerobic conditions is 74 days.

Ecotoxicity: Acetic acid is not acutely toxic to fish or invertebrates.

Prevent, by any means available, spillage from entering drains or water courses.

**DO NOT** discharge into sewer or waterways.

## Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
water	LOW	LOW
acetic acid glacial	LOW	LOW
ammonium thiocyanate	LOW	LOW
methanol	LOW	LOW
ammonium ferrous sulfate	HIGH	HIGH
carbohydrazide	HIGH	HIGH

## Bioaccumulative potential

Ingredient	Bioaccumulation
water	LOW (LogKOW = -1.38)
acetic acid glacial	LOW (LogKOW = -0.17)
ammonium thiocyanate	LOW (LogKOW = 0.5829)
methanol	LOW (BCF = 10)
ammonium ferrous sulfate	LOW (LogKOW = -2.2002)
carbohydrazide	LOW (LogKOW = -3.7307)

## Mobility in soil

Continued...

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

Ingredient	Mobility
water	LOW (KOC = 14.3)
acetic acid glacial	HIGH (KOC = 1)
ammonium thiocyanate	LOW (KOC = 4.5)
methanol	HIGH (KOC = 1)
ammonium ferrous sulfate	LOW (KOC = 6.124)
carbohydrazide	LOW (KOC = 15.65)

### SECTION 13 DISPOSAL CONSIDERATIONS

#### Waste treatment methods

Product / Packaging disposal	Dispose of according to federal, state, and local regulations. Dispose of according to federal, state, and local regulations.
------------------------------	--

### SECTION 14 TRANSPORT INFORMATION

#### Labels Required

	
Marine Pollutant	NO

#### Land transport (DOT)

UN number	1760
UN proper shipping name	Corrosive liquids, n.o.s. (contains sulfuric acid and acetic acid glacial)
Transport hazard class(es)	Class 8 Subrisk Not Applicable
Packing group	II
Environmental hazard	Not Applicable
Special precautions for user	Hazard Label 8 Special provisions B2, IB2, T11, TP2, TP27

#### Air transport (ICAO-IATA / DGR)

UN number	1760
UN proper shipping name	Corrosive liquid, n.o.s. * (contains sulfuric acid and acetic acid glacial)
Transport hazard class(es)	ICAO/IATA Class 8 ICAO / IATA Subrisk Not Applicable ERG Code 8L
Packing group	II
Environmental hazard	Not Applicable
Special precautions for user	Special provisions A3 A803 Cargo Only Packing Instructions 855 Cargo Only Maximum Qty / Pack 30 L Passenger and Cargo Packing Instructions 851 Passenger and Cargo Maximum Qty / Pack 1 L Passenger and Cargo Limited Quantity Packing Instructions Y840 Passenger and Cargo Limited Maximum Qty / Pack 0.5 L

#### Sea transport (IMDG-Code / GGVSee)

UN number	1760
UN proper shipping name	CORROSIVE LIQUID, N.O.S. (contains sulfuric acid and acetic acid glacial)
Transport hazard class(es)	IMDG Class 8 IMDG Subrisk Not Applicable
Packing group	II

# Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

<b>Environmental hazard</b>	Not Applicable
<b>Special precautions for user</b>	
EMS Number	F-A , S-B
Special provisions	274
Limited Quantities	1 L

## Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

## SECTION 15 REGULATORY INFORMATION

### Safety, health and environmental regulations / legislation specific for the substance or mixture

#### WATER(7732-18-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

#### ACETIC ACID GLACIAL(64-19-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - California Permissible Exposure Limits for Chemical Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

US - Hawaii Air Contaminant Limits

Contaminants

US - Idaho - Limits for Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Massachusetts - Right To Know Listed Chemicals

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US - Michigan Exposure Limits for Air Contaminants

US ACGIH Threshold Limit Values (TLV)

US - Minnesota Permissible Exposure Limits (PELs)

US CWA (Clean Water Act) - List of Hazardous Substances

US - Oregon Permissible Exposure Limits (Z-1)

US NIOSH Recommended Exposure Limits (RELS)

US - Pennsylvania - Hazardous Substance List

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US - Rhode Island Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US TSCA Chemical Substance Inventory - Interim List of Active Substances

#### AMMONIUM THIOCYANATE(1762-95-4) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Massachusetts - Right To Know Listed Chemicals

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US - Pennsylvania - Hazardous Substance List

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US CWA (Clean Water Act) - List of Hazardous Substances

#### METHANOL(67-56-1) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - California - Proposition 65 - Priority List for the Development of MADLs for Chemicals Causing Reproductive Toxicity

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELS)

Contaminants

US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)

US - Washington Permissible exposure limits of air contaminants

US - California Permissible Exposure Limits for Chemical Contaminants

US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants

US - California Proposition 65 - Reproductive Toxicity

US ACGIH Threshold Limit Values (TLV)

US - Hawaii Air Contaminant Limits

US Clean Air Act - Hazardous Air Pollutants

US - Idaho - Limits for Air Contaminants

US EPCRA Section 313 Chemical List

US - Massachusetts - Right To Know Listed Chemicals

US NIOSH Recommended Exposure Limits (RELS)

US - Michigan Exposure Limits for Air Contaminants

US Office of Environmental Health Hazard Assessment Proposition 65 No Significant Risk

US - Minnesota Permissible Exposure Limits (PELs)

Levels (NSRLs) for Carcinogens and Maximum Allowable Dose Levels (MADLs) for

US - Oregon Permissible Exposure Limits (Z-1)

Chemicals Causing Reproductive Toxicity

US - Pennsylvania - Hazardous Substance List

US OSHA Permissible Exposure Levels (PELs) - Table Z1

US - Rhode Island Hazardous Substance List

US Spacecraft Maximum Allowable Concentrations (SMACs) for Airborne Contaminants

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

#### AMMONIUM FERROUS SULFATE(7783-85-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

US - Alaska Limits for Air Contaminants

US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants

US - California Permissible Exposure Limits for Chemical Contaminants

US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants

US - Hawaii Air Contaminant Limits

US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air

US - Massachusetts - Right To Know Listed Chemicals

Contaminants

US - Michigan Exposure Limits for Air Contaminants

US - Washington Permissible exposure limits of air contaminants

US - Minnesota Permissible Exposure Limits (PELs)

US ACGIH Threshold Limit Values (TLV)

US - Oregon Permissible Exposure Limits (Z-1)

US CWA (Clean Water Act) - List of Hazardous Substances

US - Pennsylvania - Hazardous Substance List

US List of Active Substances Exempt from the TSCA Inventory Notifications (Active-Inactive) Rule

US - Rhode Island Hazardous Substance List

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

#### SULFURIC ACID(7664-93-9) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Continued...

## Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs	US - Vermont Permissible Exposure Limits Table Z-1-A Final Rule Limits for Air Contaminants
International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Passenger and Cargo Aircraft	US - Vermont Permissible Exposure Limits Table Z-1-A Transitional Limits for Air Contaminants
US - Alaska Limits for Air Contaminants	US - Washington Permissible exposure limits of air contaminants
US - California OEHHA/ARB - Acute Reference Exposure Levels and Target Organs (RELs)	US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values
US - California OEHHA/ARB - Chronic Reference Exposure Levels and Target Organs (CRELs)	US - Wyoming Toxic and Hazardous Substances Table Z1 Limits for Air Contaminants
US - California Permissible Exposure Limits for Chemical Contaminants	US ACGIH Threshold Limit Values (TLV)
US - Hawaii Air Contaminant Limits	US ACGIH Threshold Limit Values (TLV) - Carcinogens
US - Idaho - Limits for Air Contaminants	US CWA (Clean Water Act) - List of Hazardous Substances
US - Massachusetts - Right To Know Listed Chemicals	US Drug Enforcement Administration (DEA) List I and II Regulated Chemicals
US - Michigan Exposure Limits for Air Contaminants	US EPCRA Section 313 Chemical List
US - Minnesota Permissible Exposure Limits (PELs)	US National Toxicology Program (NTP) 14th Report Part A Known to be Human Carcinogens
US - New Jersey Right to Know - Special Health Hazard Substance List (SHHSL): Carcinogens	US NIOSH Recommended Exposure Limits (RELs)
US - Oregon Permissible Exposure Limits (Z-1)	US OSHA Permissible Exposure Levels (PELs) - Table Z1
US - Pennsylvania - Hazardous Substance List	US SARA Section 302 Extremely Hazardous Substances
US - Rhode Island Hazardous Substance List	US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory
US - Tennessee Occupational Exposure Limits - Limits For Air Contaminants	US TSCA Chemical Substance Inventory - Interim List of Active Substances

### **CARBOHYDRAZIDE(497-18-7) IS FOUND ON THE FOLLOWING REGULATORY LISTS**

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	US TSCA Chemical Substance Inventory - Interim List of Active Substances
---	--

### **Federal Regulations**

#### **Superfund Amendments and Reauthorization Act of 1986 (SARA)**

##### **SECTION 311/312 HAZARD CATEGORIES**

Flammable (Gases, Aerosols, Liquids, or Solids)	No
Gas under pressure	No
Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	Yes
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	No
Reproductive toxicity	No
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	No
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	No
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No

##### **US. EPA CERCLA HAZARDOUS SUBSTANCES AND REPORTABLE QUANTITIES (40 CFR 302.4)**

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
Acetic acid	5000	2270
Ammonium thiocyanate	5000	2270
Methanol	5000	2270
Ferrous ammonium sulfate	1000	454
Sulfuric acid	1000	454

### **State Regulations**

#### **US. CALIFORNIA PROPOSITION 65**

WARNING: This product contains a chemical known to the State of California to cause cancer and birth defects or other reproductive harm

#### **US - CALIFORNIA PROPOSITION 65 - CARCINOGENS & REPRODUCTIVE TOXICITY (CRT): LISTED SUBSTANCE**

Methanol Listed

### **National Inventory Status**

Continued...

**Hydrogen Peroxide CHEMets & VACUettes Refills and Vacu-vials Ampoules, Sodium Persulfate CHEMets Refill**

National Inventory	Status
Australia - AICS	Y
Canada - DSL	Y
Canada - NDSL	N (methanol; ammonium ferrous sulfate; water; ammonium thiocyanate; acetic acid glacial; sulfuric acid; carbohydrazide)
China - IECSC	Y
Europe - EINEC / ELINCS / NLP	Y
Japan - ENCS	Y
Korea - KECL	Y
New Zealand - NZIoC	Y
Philippines - PICCS	Y
USA - TSCA	Y
<b>Legend:</b>	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing (see specific ingredients in brackets)

## SECTION 16 OTHER INFORMATION

Revision Date	13/08/2018
Initial Date	13/11/2014

### Other information

#### Ingredients with multiple cas numbers

Name	CAS No
ammonium ferrous sulfate	10045-89-3, 7783-85-9, 10414-76-3, 17685-66-4, 220710-26-9

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

PC – TWA: Permissible Concentration-Time Weighted Average  
 PC – STEL: Permissible Concentration-Short Term Exposure Limit  
 IARC: International Agency for Research on Cancer  
 ACGIH: American Conference of Governmental Industrial Hygienists  
 STEL: Short Term Exposure Limit  
 TEEL: Temporary Emergency Exposure Limit  
 IDLH: Immediately Dangerous to Life or Health Concentrations  
 OSF: Odour Safety Factor  
 NOAEL :No Observed Adverse Effect Level  
 LOAEL: Lowest Observed Adverse Effect Level  
 TLV: Threshold Limit Value  
 LOD: Limit Of Detection  
 OTV: Odour Threshold Value  
 BCF: BioConcentration Factors  
 BEI: Biological Exposure Index

Powered by AuthorITe, from Chemwatch.